

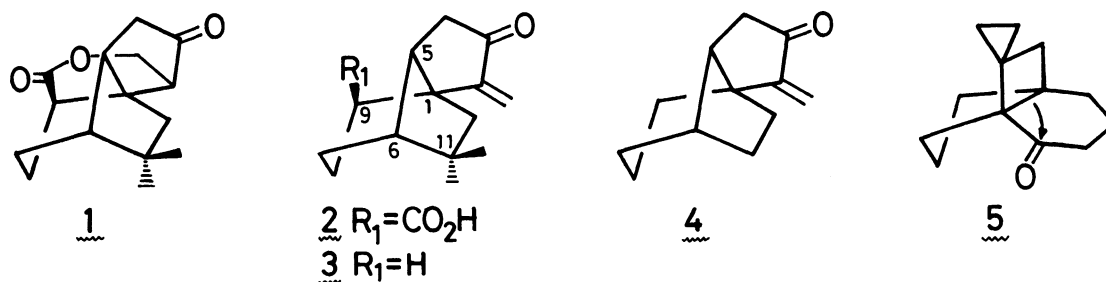
AN ALTERNATIVE SYNTHESIS OF (\pm)-DESCARBOXYQUADRONE

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The alternative synthesis of descarboxyquadrone by using the acid-catalyzed rearrangement of [4.3.2]propellane derivative and its antitumor activity are described.

Quadrone (1)¹⁾ and its retrolactonization compound, terrecyclic acid A (2),²⁾ isolated from the same fungus, have been shown to display significant biological activities involving antitumor activity. From the view point of their biological activities, descarboxyquadrone (3), having an α -methylene cyclopentanone structure and geminal methyl groups at C-11 on tricyclo[4.3.2.0^{1,5}]undecane skeleton, has been synthesized and found to possess anti HeLa activity.³⁾ On the other hand, we have already reported the novel formal synthesis of 3 by using the acid-catalyzed rearrangement of [4.3.2]propellane 5 to the quadrone skeleton.⁴⁾ Although this one-step skeleton construction was serviceable in securing the formal synthesis, we were dissatisfied with the preparation⁵⁾ of the propellane 5, derived from the minor product of photocycloaddition of the enone 6 to allene. We describe herein an alternative and highly selective sequence for the total synthesis of 3 and the interesting antitumor activity of 3 along with that of the binor derivative, 2-methylenetricyclo[4.3.2.0^{1,5}]undecan-3-one (4).⁴⁾



On the basis of our studies on the acid-catalyzed rearrangements of [m.n.2]-propellanes,⁶⁾ we planned the skeletal transformation of [4.3.2]propellane 7 to the quadrone skeleton as the initial step and then the introduction of geminal methyl groups to C-11 position as outlined in Scheme 1. To this purpose, photo-reaction of 6 [$\text{CH}_2=\text{CHOAc}$, Et_2O , -78°C , 3.5 h] was undertaken to give the propellane 7⁷⁾ in 91% yield together with a small amount (<4%) of three other cyclo-

adducts. The regiochemistry⁸⁾ of 7 was assumed to be head to tail for the sake of previous works on photocycloaddition of α,β -enones to vinyl acetate,⁹⁾ and was finally ascertained by the synthesis of 3 from 7. The acid-catalyzed rearrangement of 7 [concd HCl, Et₂O, reflux, 36 h] afforded the diol 8⁷⁾ (mp, 127-130 °C) in 72% yield whose secondary hydroxyl group was oxidized quantitatively to the keto alcohol 9 with PCC¹⁰⁾ [C₅H₅N·CrO₃·HCl, CH₂Cl₂, rt, 2.5 h]. Conversion of 9 into the keto olefin 11⁷⁾ was carried out by dehydration [SOCl₂, Py, CH₂Cl₂, rt, 4 h] followed by reduction of 10⁷⁾ with tributyltin hydride [2,2'-azobisisobutyronitrile, cyclohexane, reflux, 4 h] in 88% overall yield. The geminal methyl groups were introduced by treatment of 11 with sodium hydride and methyl iodide [5 equiv. NaH, THF, 50 °C, 2 h, then 6 equiv. CH₃I, reflux, 12 h] to yield the intermediate 12⁷⁾ in 78% yield.

With the desired compound 12 in hand, deoxygenation of carbonyl group at C-10 and then allylic oxidation at C-3 would lead the key intermediate 15 which had been synthesized by us.⁴⁾ However, attempted Wolff-Kishner reduction of 12 failed to produce the desired product 14, and, therefore, a more circuitous route for removal of the very hindered carbonyl group had to be employed. Reduction of 12 with LiAlH₄ [Et₂O, rt, 4 h] gave the alcohol 13¹¹⁾ in 98% yield as a sole product. While radical deoxygenation¹²⁾ with Li/EtNH₂ via phosphoramidate of 13 was inadequate to our aim, the Barton-McCombie reaction¹³⁾ of 13 [i) NaH, CS₂, imidazole, HMPA, THF, reflux, 22 h, then Me₂SO₄, reflux, 1 h; ii) *n*-Bu₃SnH, AIBN, toluene, reflux, 13 h] served admirably to give the olefin 14⁷⁾ in 33% overall yield. Allylic oxidation¹⁴⁾ of 14 [CrO₃, 3,5-dimethylpyrazole, CH₂Cl₂, -25 ~ -20 °C, 5 h] afforded the known enone 15⁴⁾ in 60% yield.

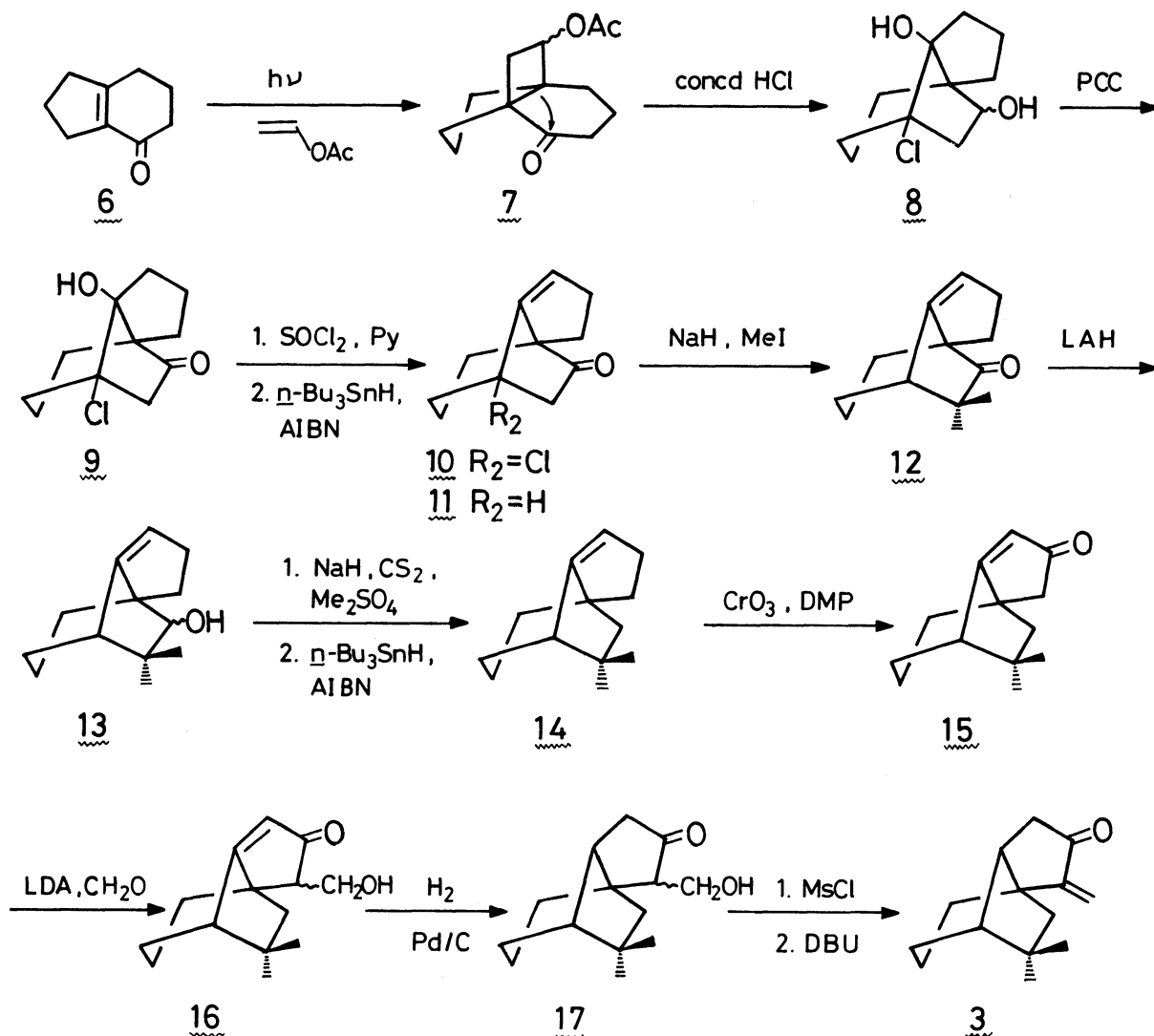
At the final stage, α -methylenation of 15 to descarboxyquadrone (3) was accomplished via almost the same procedure as that of Smith et al.³⁾ Namely, hydroxymethylation of 15 [1.1 equiv. LDA, THF, -78 °C, then CH₂O, -20 °C, 25 min] and subsequent hydrogenation of 16 [Pd/C, AcOEt, rt, 20 h] followed by dehydration of 17 [i) MeSO₂Cl, Py, rt, 7 h; ii) 1,8-diazabicyclo[5.4.0]undec-7-ene, C₆H₆, rt, 14 h] furnished 3^{3,15)} in 52% overall yield. Thus, the total synthesis of 3 has been achieved in 14 steps and 4.6% overall yield based on the bicyclic enone 6.

The bioassay of 3 and its binor derivative 4⁴⁾ without geminal methyl groups at C-11 in 3 was undertaken against tumor cells of mice in vitro and the results are summarized in Table 1 together with those of natural quadrone (1). Interestingly, 3 and 4 exhibited higher cytotoxicity than the antibiotic 1. Therefore, it is reasonable to conclude that the α -methylene cyclopentanone

Table 1. Antitumor Activity of Quadrone (1), Descarboxyquadrone (3), and the Related Compound 4

Test cell	IC ₅₀ , ng/ml		
	<u>1</u>	<u>3</u>	<u>4</u>
P388	190	136	94
L1210	650	64	185

structure on the tricyclic skeleton plays more important role in the appearance of antitumor activity of quadrone and the related compounds than the carboxyl and geminal methyl groups.



Scheme 1.

We would like to thank Dr. N. Ida and Mr. H. Koike of Basic Laboratories, Toray Industries, Inc. for the screening.

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 - 7) All new compounds isolated gave satisfactory spectral and analytical data. Selected data are as follows:
 - 7: IR (neat) 1730, 1690, 1230, 1040 cm^{-1} ; ^1H NMR (CCl_4) δ 1.1-2.6 (m, 17H containing s at 2.00), 4.64 (t, $J = 7$ Hz, 1H).
 - 8: IR (KBr) 3400, 1040, 1010 cm^{-1} ; ^1H NMR (CDCl_3) δ 1.2-2.6 (m, 16H), 3.72-3.94 (m, 1H).
 - 12: IR (neat) 3050, 1740, 1680, 1380 cm^{-1} ; ^1H NMR (CCl_4) δ 0.96 (s, 3H), 1.11 (s, 3H), 1.4-1.8 (m, 8H), 2.4-2.7 (m, 3H), 5.20 (t, $J = 2$ Hz, 1H).
 - 14: IR (neat) 3050, 1680, 1380 cm^{-1} ; ^1H NMR (CCl_4) δ 0.94 (s, 3H), 1.12 (s, 3H), 1.2-1.8 (m, 10H), 1.93 (m, 1H), 2.36-2.62 (m, 2H), 4.96 (t, $J = 2$ Hz, 1H).
 - 8) The configuration of the acetoxyl group of 7 was not determined.
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 - 15) ^{13}C NMR data not reported in ref. 3 are as follows: (CDCl_3) δ 208.2 (s), 154.1 (s), 114.2 (t), 54.0 (d), 53.6 (s), 51.4 (t), 49.3 (d), 41.8 (t), 41.6 (s), 36.8 (t), 35.1 (q), 31.0 (t), 27.1 (q), 19.5 (t).

(Received January 16, 1986)